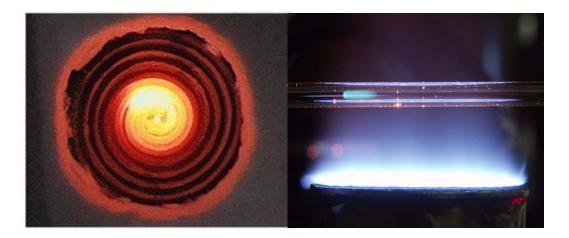
The First International Workshop on Flame Chemistry, July 28 - 29, 2012, Warsaw, Poland



Micro flow reactor with prescribed temperature profile

Toward fuel Indexing and kinetics study based on multiple weak flames

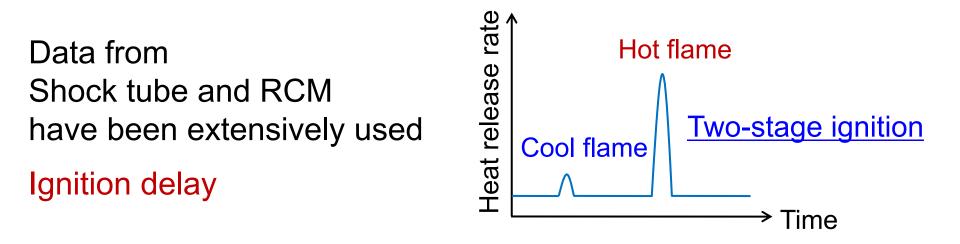
Kaoru Maruta, Tohoku Univ.

S. Minaev, N.I. Kim, T. Yokomori, H. Nakamura, S. Hasegawa, T. Tezuka T. Kataoka, Y. Tsuboi, H. Oshibe, A. Yamamoto, R. Tanimoto, M. Hori, K. Saruwatari, S. Suzuki, T. Kamada, X. Li, Y. Kizaki, T. Onishi and H. Takahashi



Background and objectives

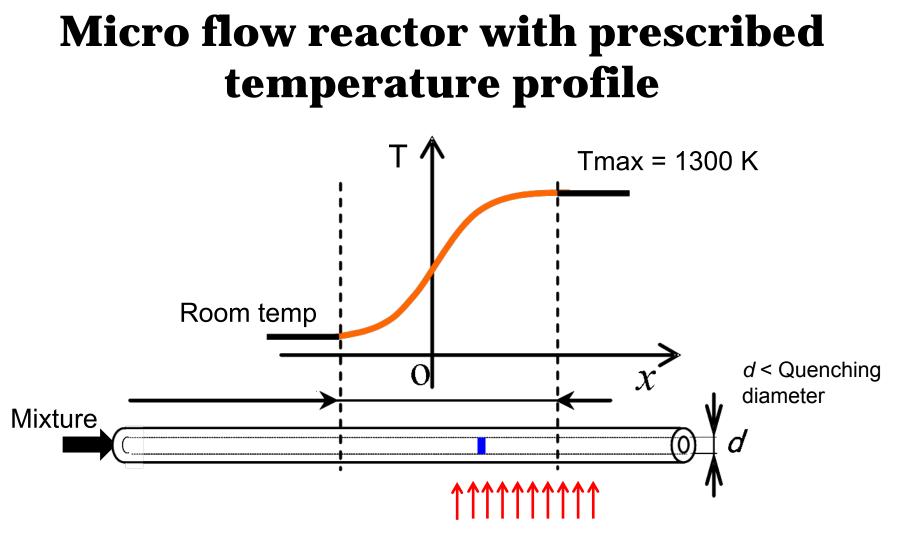
For understandings ignition and combustion characteristics of practical fuels...



Micro flow reactor with prescribed temperature profile

Single or multiple weak flames

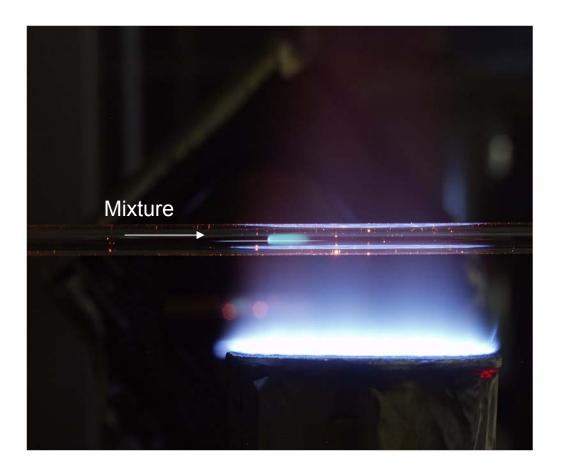


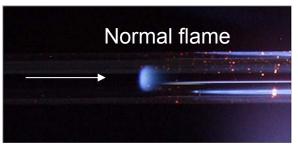


тоноки

Stationary wall-temperature profile by an external heat source Inner diameter of the tube < conventional quenching diameter Gas phase temperature governed by wall temperature profile Laminar flow and constant pressure

Flame behaviors in a micro flow reactor with a prescribed temperature profile

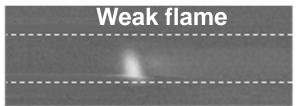




High velocity region



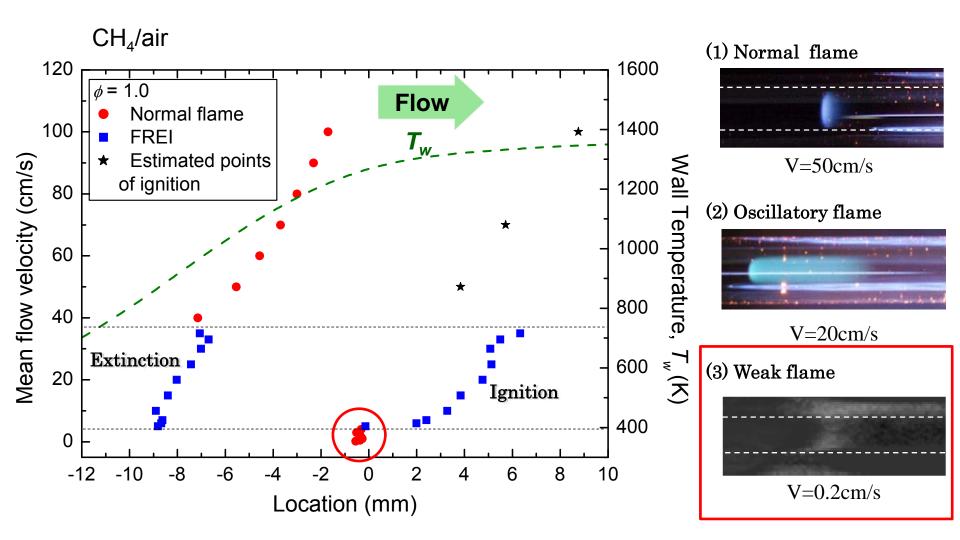
Intermediate velocity region



Low velocity region



Three kinds of flame responses

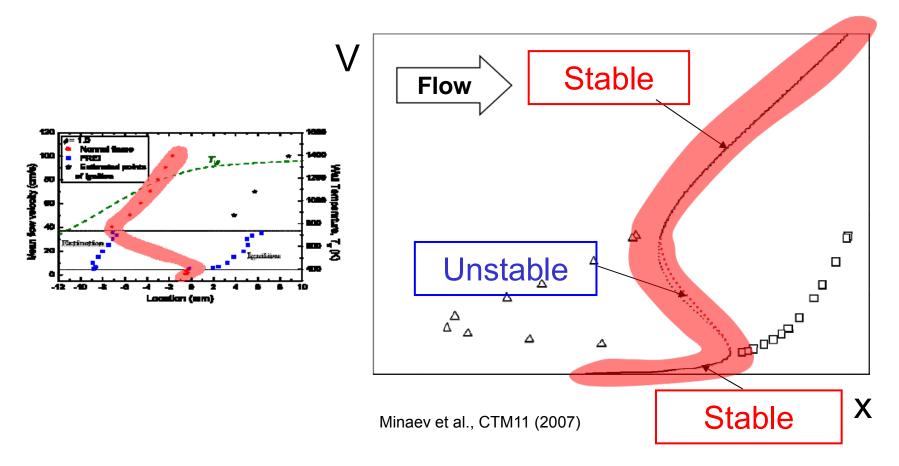




Normal flame, oscillatory flame, weak flame

Maruta et al., PCI30 (2005)

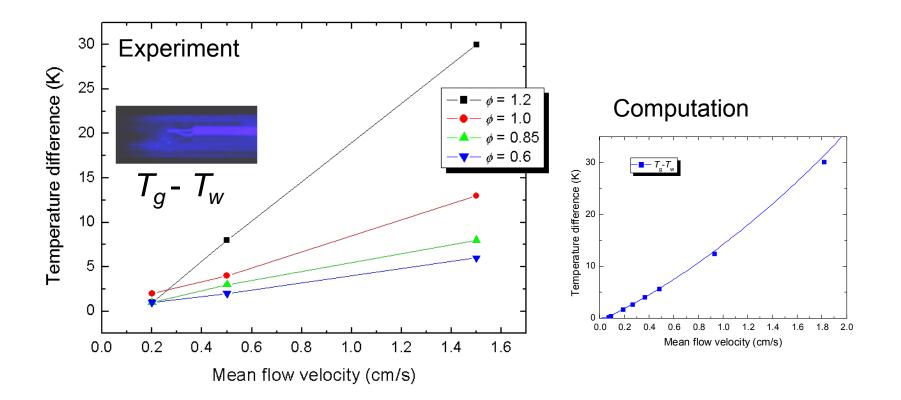
Theoretical S-shaped response



Two stable and one unstable solutions predicted theoretically → Weak flame corresponds to ignition branch



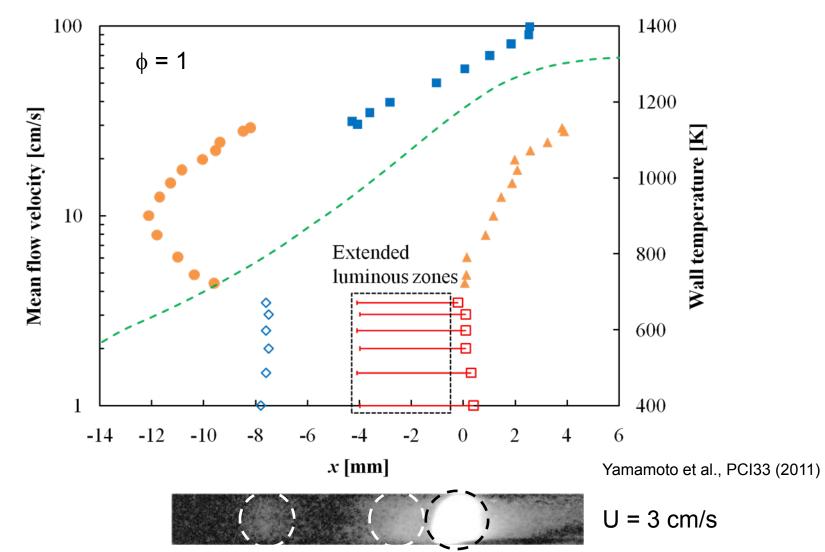
Lower limit of weak flames identified



At V = 0.2 cm/s, $T_w = 1225$ K, $T_g - T_w < 2$ K for CH_4 /air mixture Extremely small temperature increase near lower limit Flame position close to the ignition limit Weak flame location \rightarrow Ignition temperature



Triple weak flames, n-heptane





Triple stationary weak flames observed Weak flame location (temp.) insensitive to flow velocity

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Computations (one-dimensional plug flow)

<u>Code</u> PREMIX with small modification

Gas-phase energy equation

$$\dot{M}\frac{dT}{dx} - \frac{1}{c_p}\frac{d}{dx}\left(\lambda A\frac{dT}{dx}\right) + \frac{A}{c_p}\sum_{k=1}^{K}\rho Y_k V_k c_{pk}\frac{dT}{dx} + \frac{A}{c_p}\sum_{k=1}^{K}\dot{\omega}_k h_k W_k \left[-\frac{A}{c_p}\frac{4\lambda Nu}{d^2}(T_w - T_v)\right] = 0$$

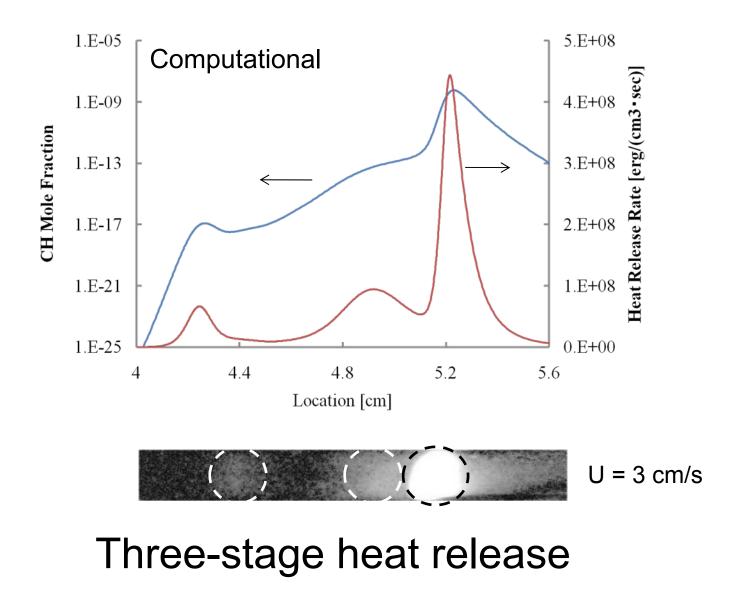
Heat transfer with the wal

Reaction scheme*n*-heptane, reduced mechanism from LLNL(159 species, 1540 steps)Seiser et al., PCI 28 (2000)

<u>Conditions</u> Stoichiometric gaseous *n*-heptane/air mixture Experimental wall temperature profile was provided as Tw(x)

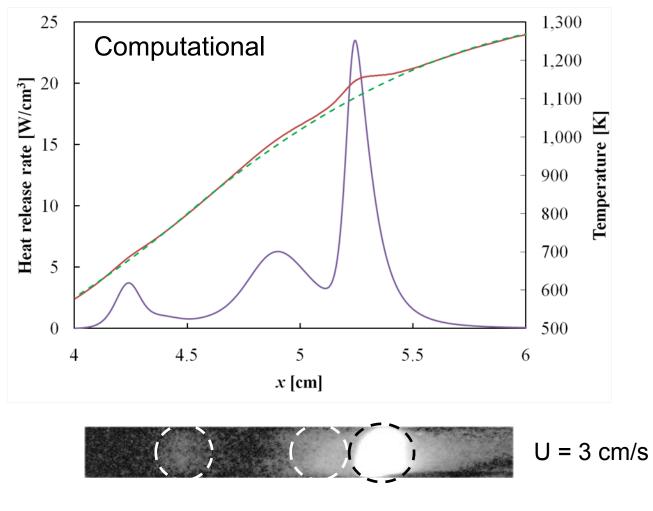
Flame position Peaks of heat-release-rate (HRR) [W/cm³] profile

Triple weak flames, n-heptane





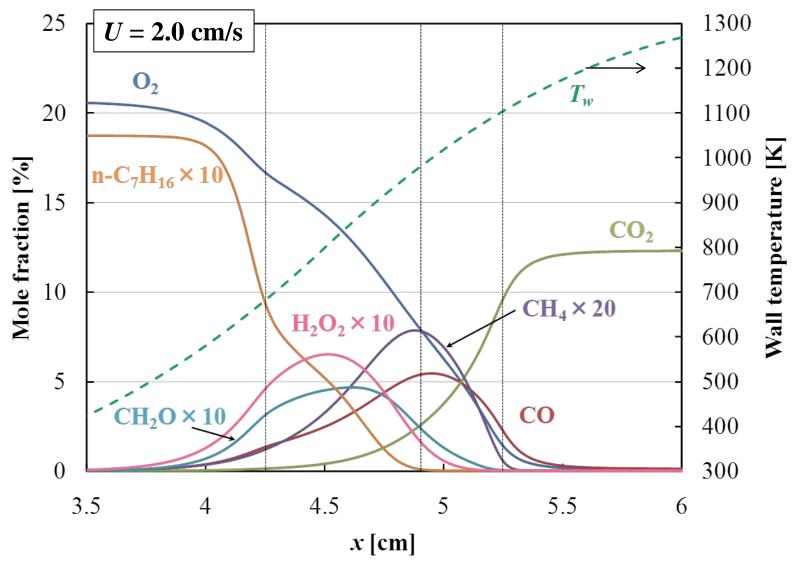
Triple weak flames, n-heptane





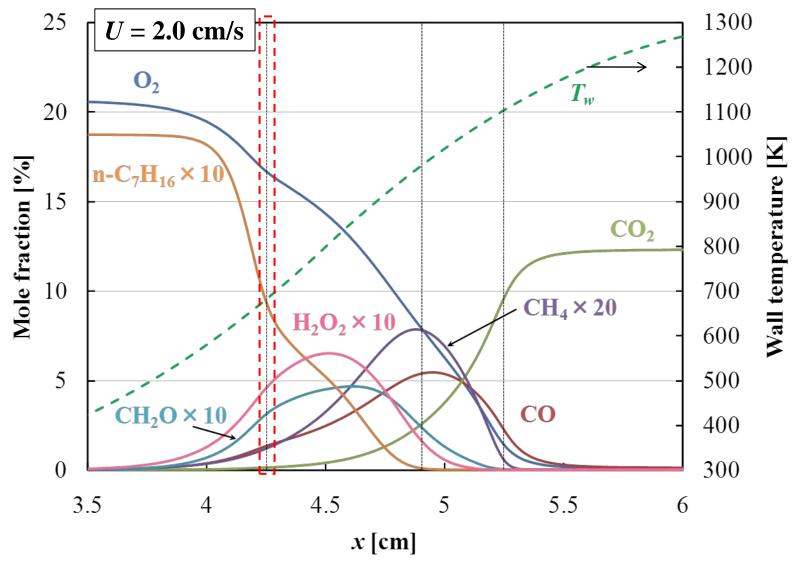
Three-stage heat release

Computational species profiles





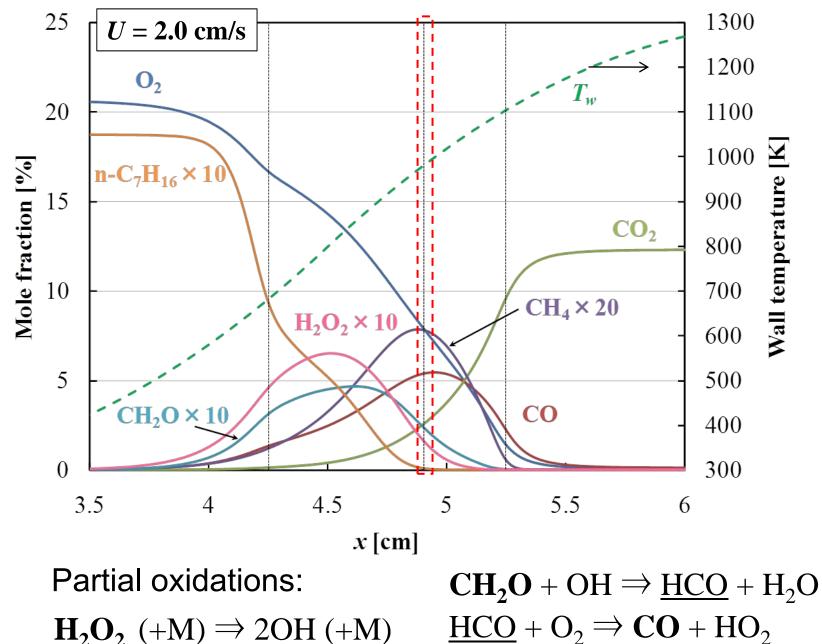
Computational species profiles -1st weak flame-





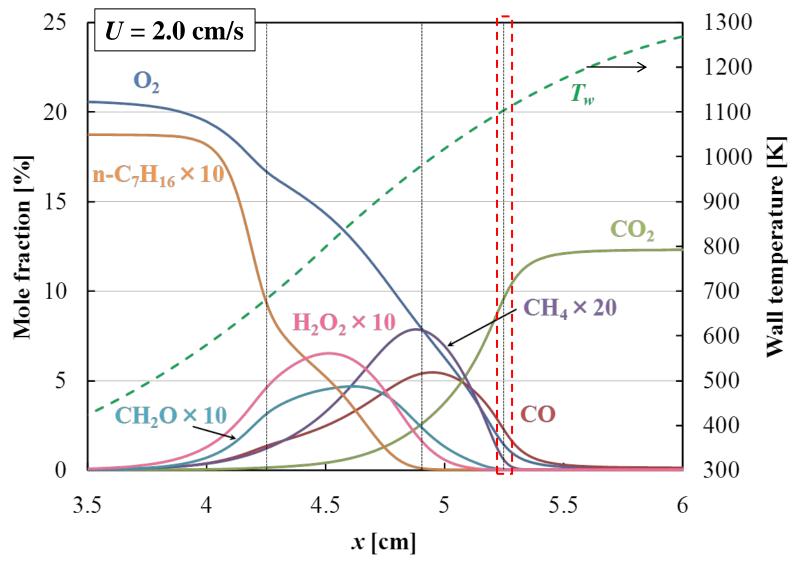
LTO: CH₂O, H₂O₂, CO, CH₄ produced

Computational species profiles -2nd weak flame-





Computational species profiles -3rd weak flame-

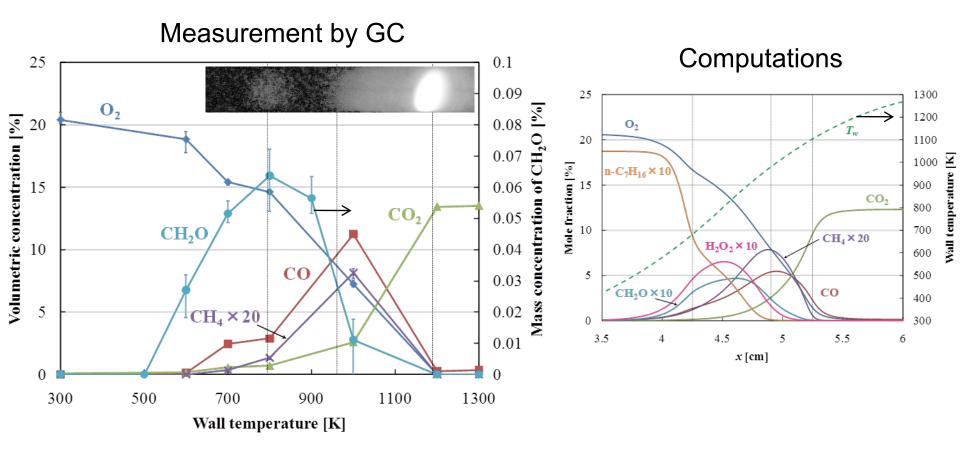




Full oxidations: $CO + OH \Rightarrow CO_2 + H$

Comparison: measurements and computations

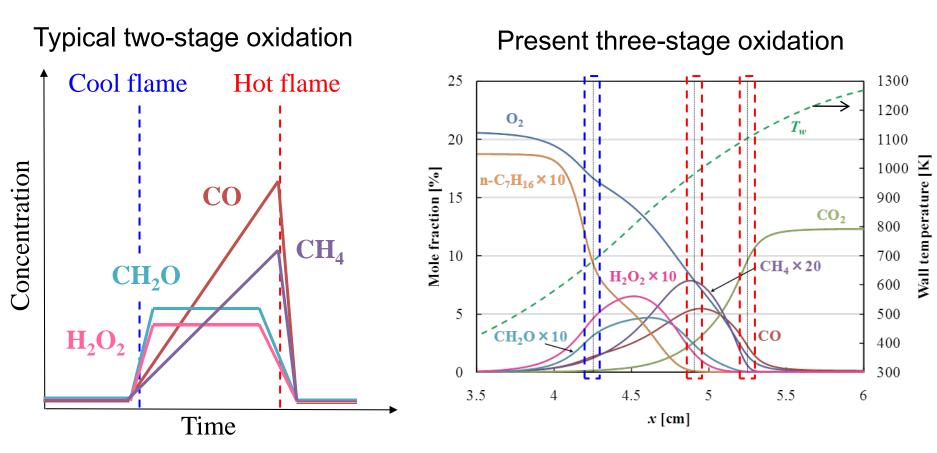
(U = 2.0 cm/s)



Three-stage oxidation process was experimentally confirmed by gas sampling



Interpretation of triple weak flames in MFR



Typical two-stage oxidation: Cool flame + Hot flame Three-stage oxidation: Cool flame + Separated hot flames (Blue flame & Hot flame)



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MFR applied for gasoline PRF

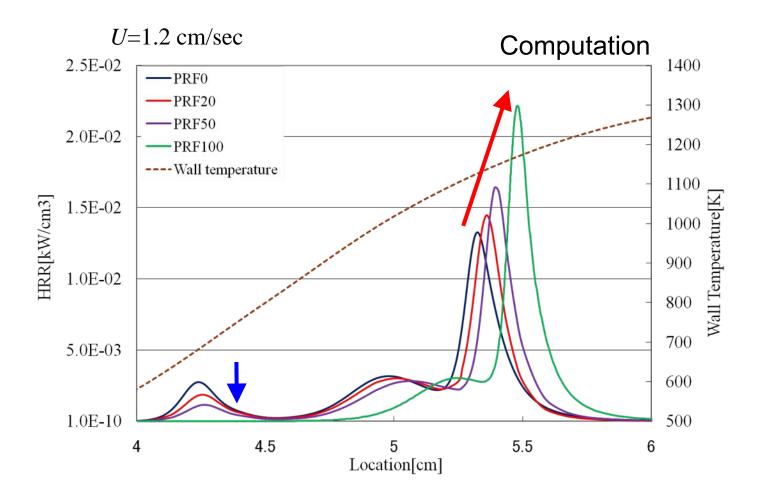
n-heptane + *iso*-octane (PRF)

PRFO	
PRF20	
PRF50	
PRF100	

Appearances of multiple weak flame represent Research Octane Number



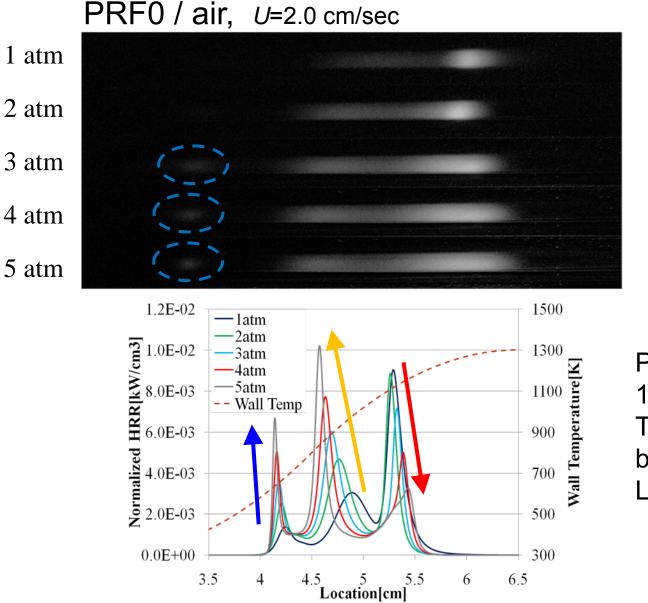
Weak flames at different RON







Weak flames at elevated pressures



PRF20, 50, 100 similar Tendencies but weaker LTO



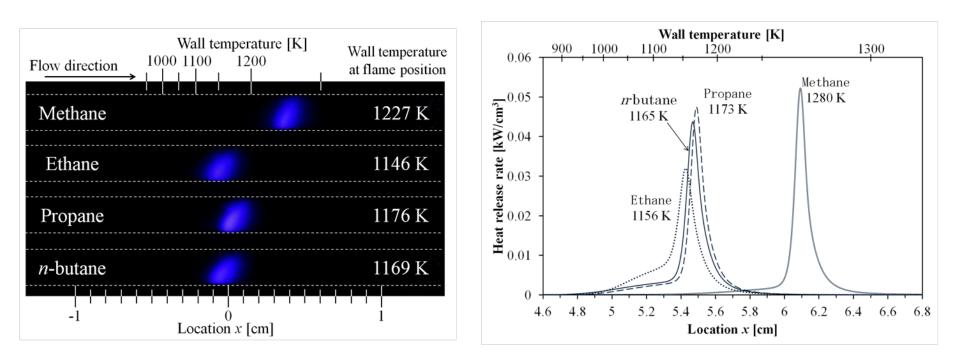
Hori et al., CNF159 (2012)

Fuels addressed

```
methane (CH_4)
DME (CH<sub>3</sub>OCH<sub>3</sub>)
n-heptane (C_7H_{16})
iso-octane (C_8H_{16})
toluene (C_7H_8 or C_6H_5CH_3)
methane (CH_4)
ethane (C_2H_6)
propane (C_3H_8)
n-butane, iso-butane (C_4H_{10}) \rightarrow (Kamada et al., WIPP)
```



Natural gas components (Kamada et al., WIPP)

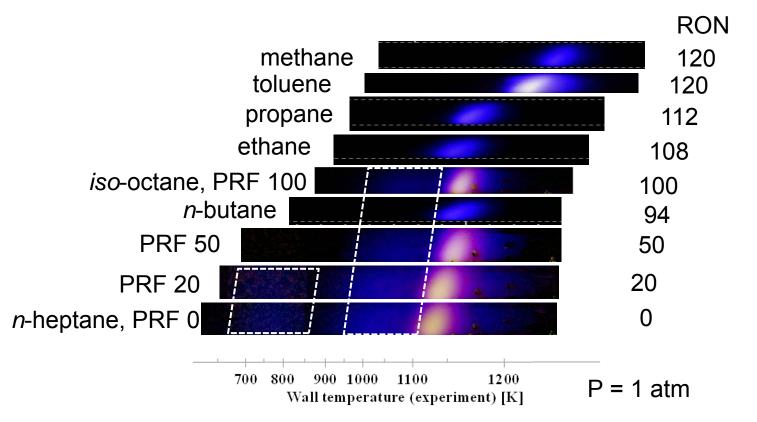


Similar single weak flame (hot flame) observed for each fuel Different flame locations Computations reproduced experimental observation



Weak flames at various RON

Data collected at slightly different conditions, i.e., flow velocity, exposure time, temperature profile





3rd weak flame (main flame) location (temp.) monotonic function of RON 2nd weak flame (blue flame) observable when RON < 100 1st weak flame (cool flame) observable when RON < 20

Conclusions and future, 1 of 2

Micro flow reactor with prescribed temperature profile was introduced

- -Three kinds of flame response (S-shaped)
- -Weak flame corresponds to ignition branch
- -Multiple weak flame utilized for fuel characterization
- -Multiple weak flames at elevated pressures
- -Appearances of weak flame correlated with RON

Conclusions and future, 2 of 2

Diesel fuels and Cetane numbers

n-cetane (hexadecane, $C_{16}H_{34}$) *iso*-cetane (2,2,4,4,6,8,8-heptamethylnonane, $C_{16}H_{34}$) *n*-decane ($C_{10}H_{22}$) α -methylnaphthalene ($C_{11}H_{10}$) \rightarrow (Suzuki et al., 5E06)

Ethanol \rightarrow (Nakamura et al., 1E02) Syngas Oxyfuel combustion \rightarrow (Li et al., WIPP) Effect of surface reaction \rightarrow (Kizaki et al, WIPP) PAH and soot

Optical diagnosis

LIF, CRDS for precise species profile measurements Higher pressures

High pressure chamber (up to 20 bar) fabricated



Acknowledgements

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Micro flow reactor now commercially available